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Red Complete the following sentences by filling in the appropriate word from the list below. Gas Liquid Solid Plasma Matter Energy Physical Chemical Evaporation Matter is anything that has mass and volume. 1. The two states of matter that occupy a definite volume are solid and liquid. 2. gas and plasma are the other two states of matter.

### **4. 2\20133 Review and Reinforcement - Matter - 4 23 ...**

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9)  $\text{Fe} + \text{CuSO}_4 \rightarrow$  10)  $\text{Cl}_2 + \text{MgI}_2 \rightarrow \dots$  Chemical Reactions Review 38 Terms. Lozinski55. Chapter 8: Chemical Equations and

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Reactions 31 Terms. michelle211. OTHER SETS BY THIS CREATOR. Ionic Charges or Oxidation Numbers of Transition Metals 10 Terms. Snyderorama. Ionic Charge or Oxidation Number of Representative Groups 8 Terms.

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speed, size, and shape 8. net force 9. microwelds 10. rolling 11. air resistance 12. acceleration 13. sliding 14. parachute 1. Gravity is a force that every object in the

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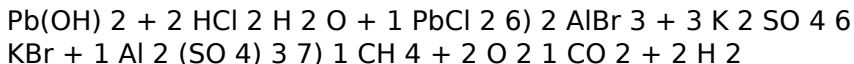
### **Chemistry 9 2 Review And Reinforcement Answers**

2-2 Explore, page 10 . 1. The thermometer can only be accurate within the range established by the melting and boiling points of water. Students may note that the expansion of water is not great enough to result in very precise calibrations. 2. Answers will vary depending on the degree of success in calibrating the thermometer.

### **2 Answer - Weebly**

Balancing Chemical Equations - Answer Key Balance the equations below: 1)  $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$  2)  $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$  3)  $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$  4)  $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$  5) 1

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There might have been an unknown influence affecting the dependent variable. (6/2) Reinforcement. Section 1 (page 11) 1-3. Accept any three of the following, or other. ... 9. h (5/2) Part B. Concept Review (page 87) 1. B (2/1) 2. U (2/1) 3. U (2/1) 4. B (2/1) 5. d (1/1) 6. c (6/2) 7. d (6/2) 8. a (3/1) 9. c (3/1) 10. a (1/1)

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11. d (6/2) 12. d ...

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