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CHAPTER 5 REVIEW The Periodic Law

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SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed

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that certain similarities in the ...

5 The Periodic Law

CH 10 Section Review 13.1 & 13.2 . CH
10 Section Review 13.3 & 13.4 . CH 10
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Matter and Change. SECTION 3. SHORT ANSWER
Answer the following questions in the space provided. 1. A horizontal row of elements in the periodic table is called a(n) period . 2. The symbol for the

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element in Period 2, Group 13, is B . 3.
Elements that are good conductors of
heat and electricity are metals .

1 Matter and Change - HUBBARD'S CHEMISTRY

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MODERN CHEMISTRY SECTION 2-3

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significant figures. a. 37.26 m 2.7 m
0.0015 m b. 256.3 mL 2 L 137 mL c. d.
5.

2 Measurements and Calculations

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3. Compare and contrast a model with a theory. Theories are broad generalizations used to explain observations. Models are a physical object used to illustrate or explain complex concepts or an explanation of how phenomena occur and how data

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and events are related. MODERN
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2 Measurements and Calculations - Kenilworth Public Schools

chapter 7, section 3 ... Holt McDougal
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Vocabulary. STUDY. PLAY. Formula

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mass. The formula unit of a molecule or ion is the sum of the average atomic masses of all atoms represented in its formula. Percentage composition. Holt McDougal Modern Chemistry Chapter 7 Section 3-4 ... Other Results for Holt Modern ...

Chapter 7 Section 3 Modern

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11. a 12. d 13. a 14. 15. a 16. d 17. c 18.
b 19. a 20. b 21. d 22. 23. c 24. b 25. a
TEST B 1. b 2. c 3. a 4. 5. b 6. d 7. d 8. c
9. decreases 10. solute 11. colloid

Assessment Chapter Test A

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